

Ecosystem effects of ocean acidification

Prof. Jason M Hall-Spencer



**PLYMOUTH
UNIVERSITY**

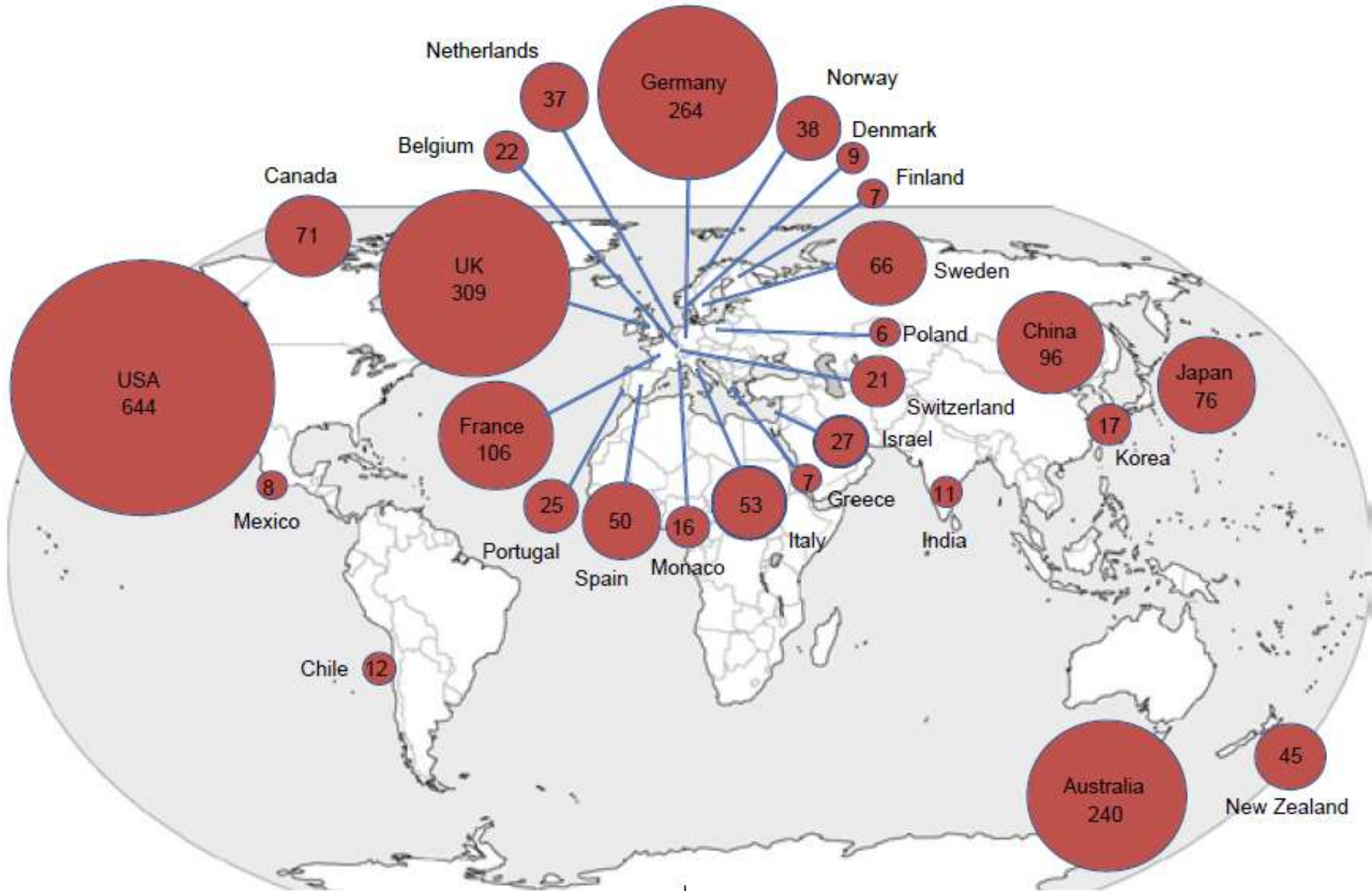
 Promoting international exchange, international
cooperation, international understanding.

THE SASAKAWA PEACE FOUNDATION

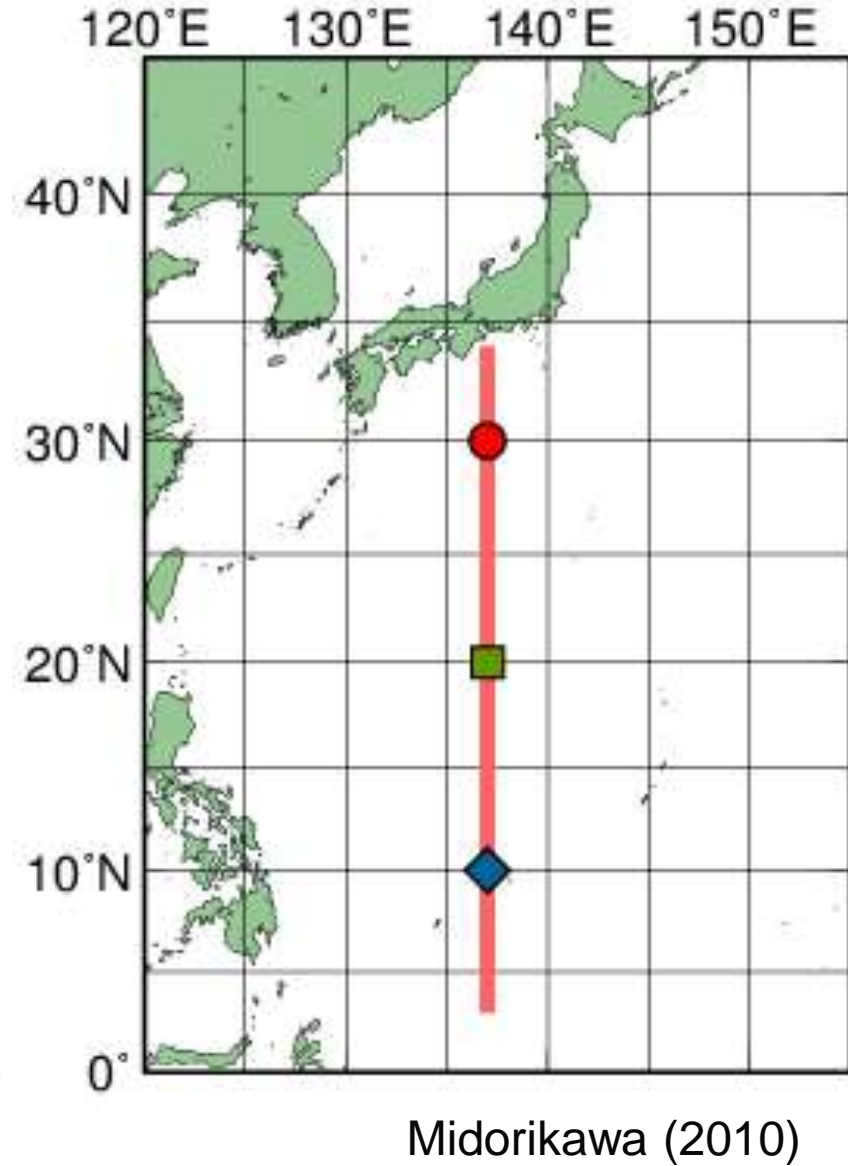
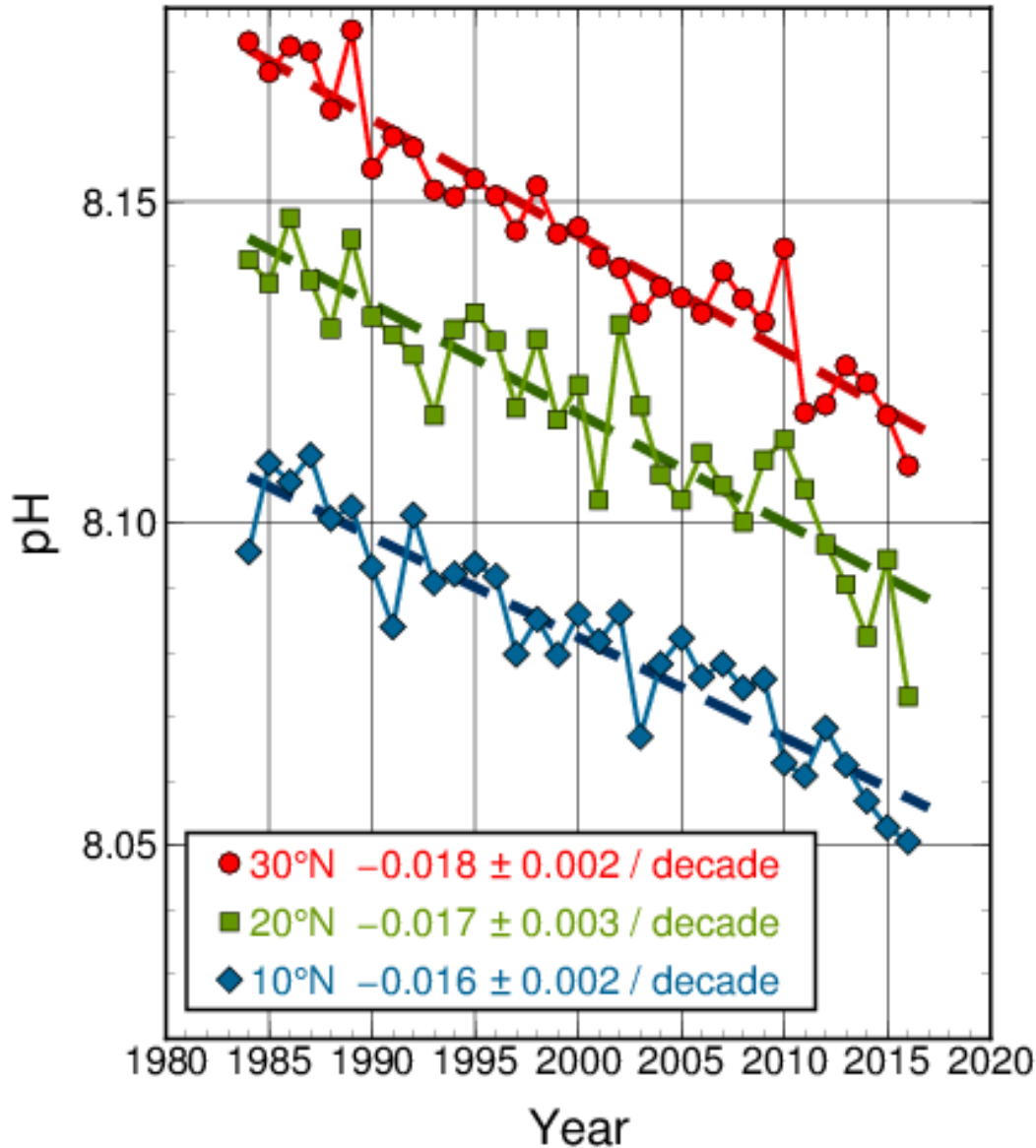
28 October 2018



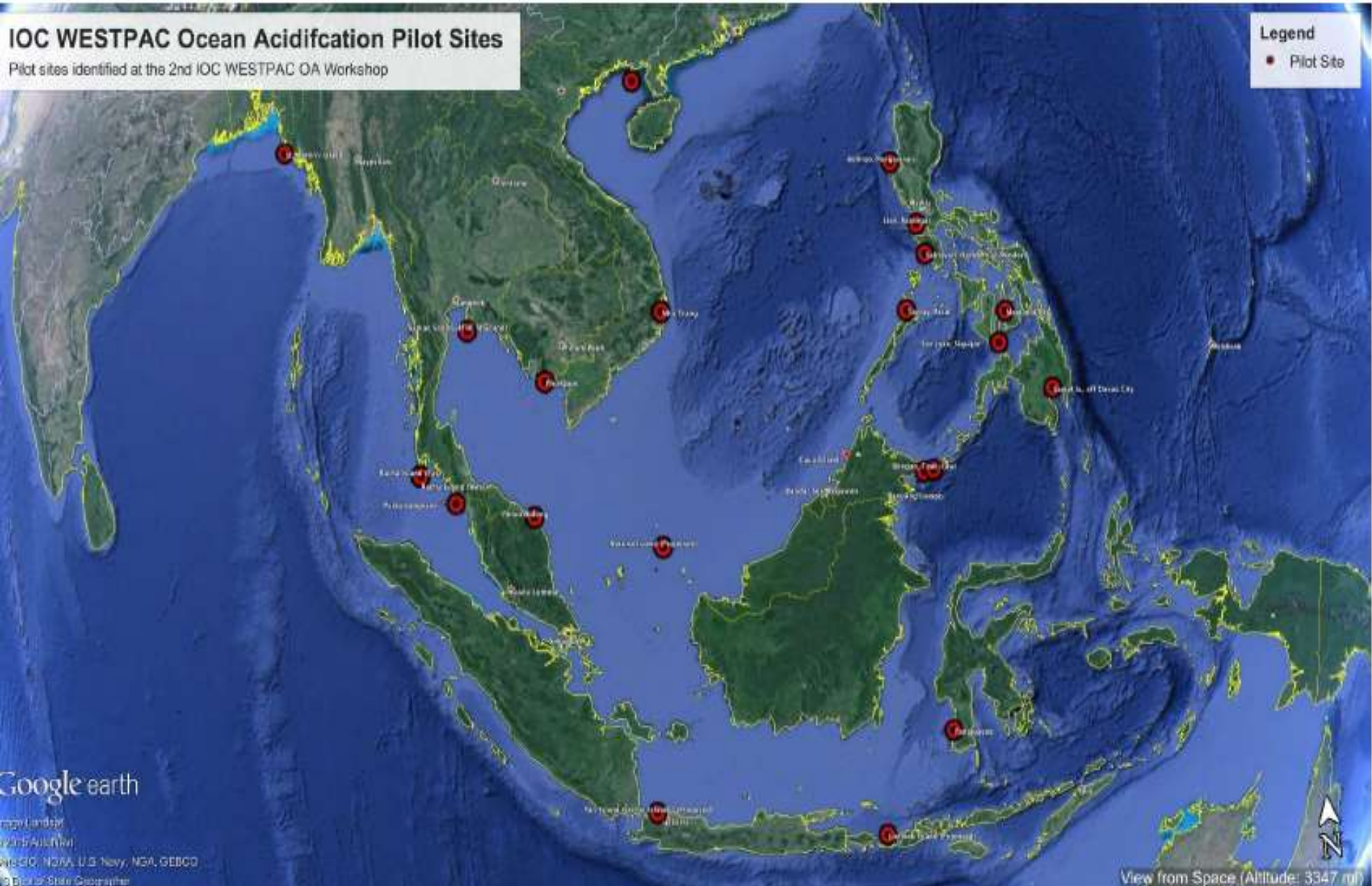
Less ocean acidification research in Asia than US, UK and Australia. Japan is a regional leader



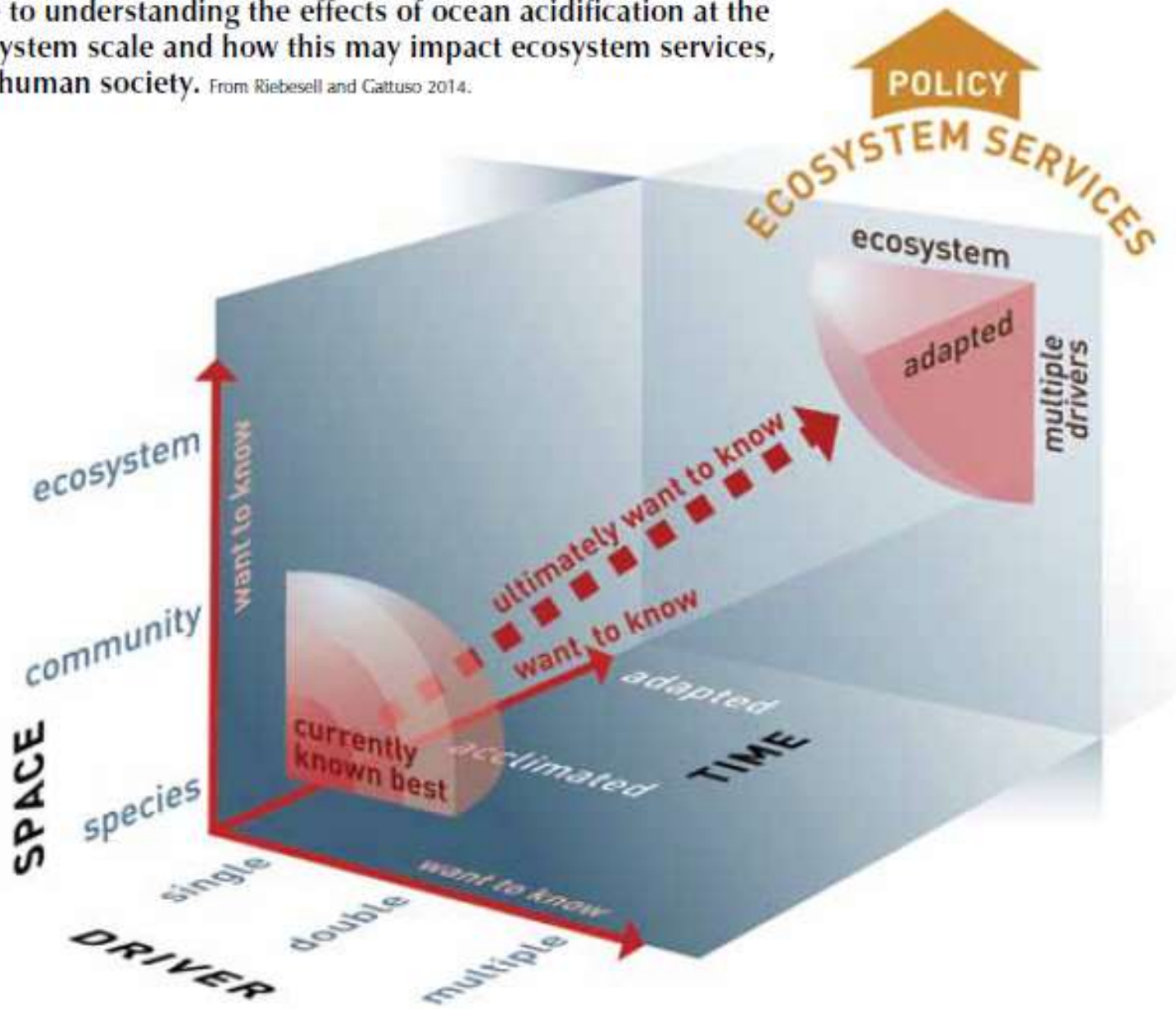
We know that seawater pH is falling rapidly off Japan



Monitoring ocean acidification is now starting in SE Asia



The scientific challenge of moving from the current knowledge base to understanding the effects of ocean acidification at the ecosystem scale and how this may impact ecosystem services, and human society. From Riebesell and Gattuso 2014.





Problem..

Last IPCC report says that we do not yet know what the ecological effects will be – as it is difficult to scale-up from laboratory experiments.



Solution..

Areas with naturally high CO₂ can help show ecosystem responses to ocean acidification

Ecosystem effects of ocean acidification and warming

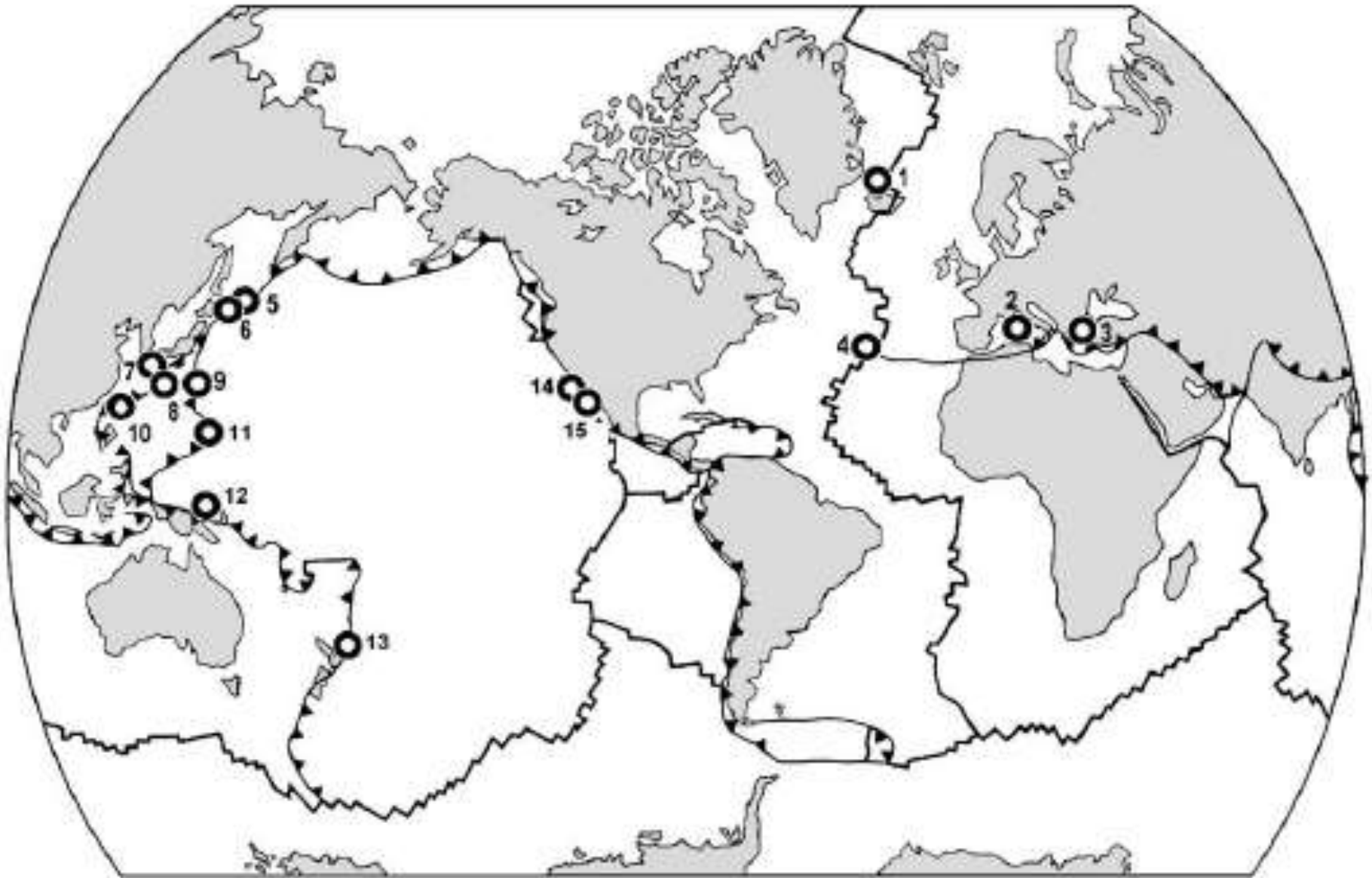
We have new data on 100s of species including Bacteria, Cyanobacteria, Diatoms, Coccolithophores, Seaweeds, Seagrasses, Sponges, Corals, Polychaetes, Crustaceans, Molluscs, Echinoderms & Fish



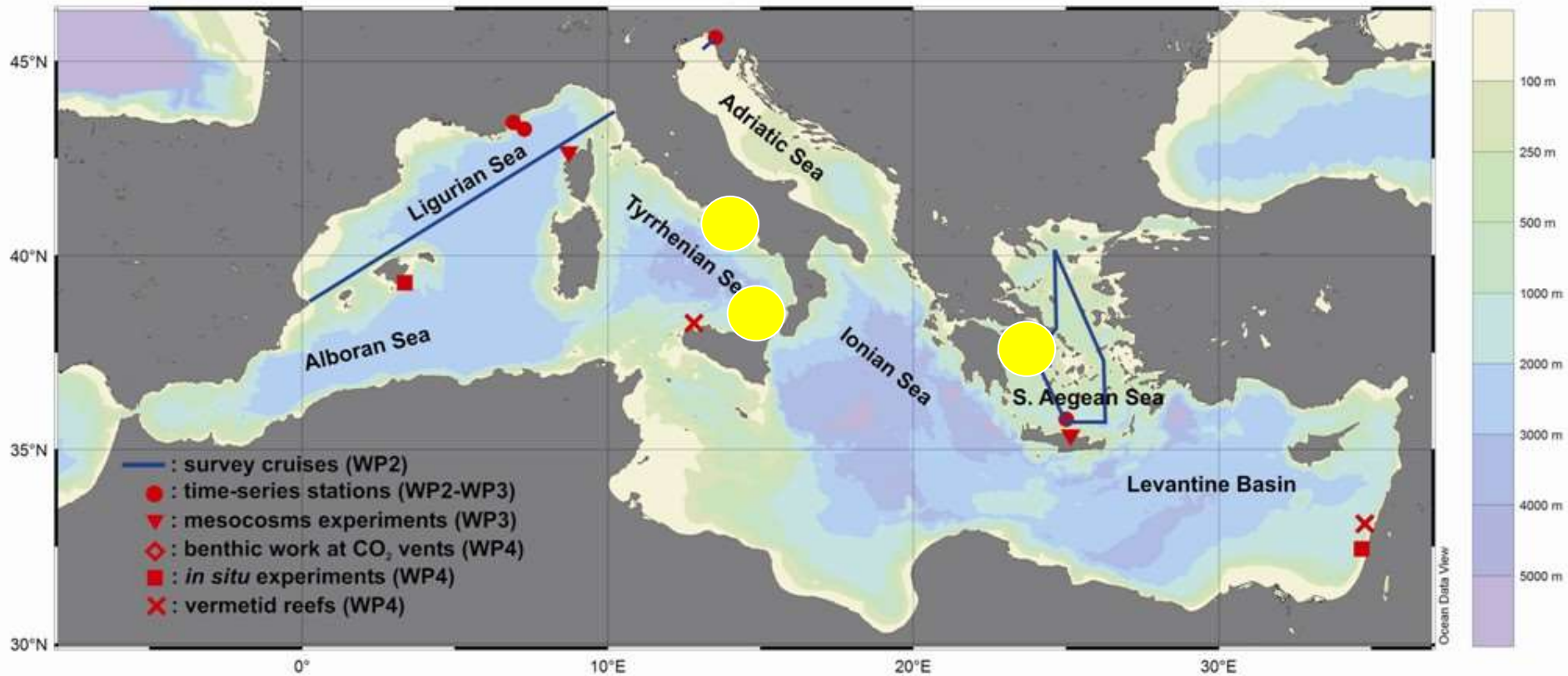


Hall-Spencer et al. 2008 Nature

Coastal CO₂ seeps worldwide



SEA CHANGE WITH PLYMOUTH UNIVERSITY



Fish reproduction is affected by acidification

Paternity tests of the eggs show that sneaker males were less successful at high CO₂ (Milazzo et al. 2016 Proc Roy Soc)



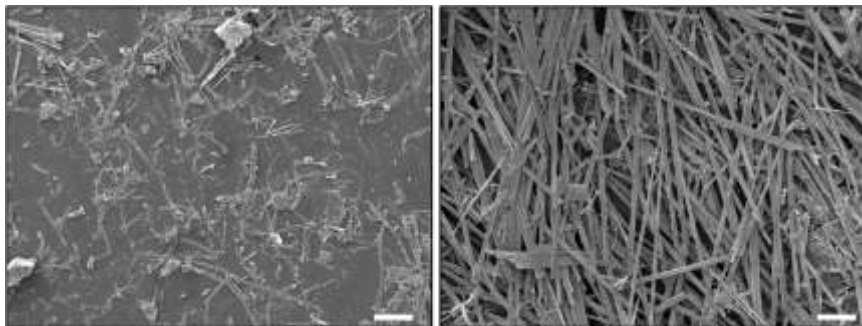
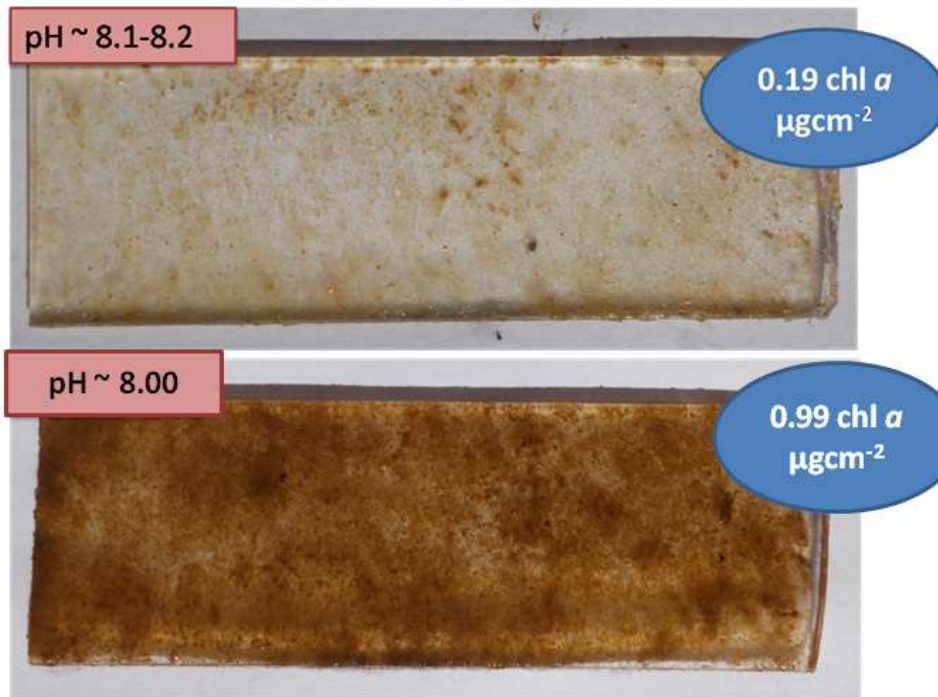
Fish reproduction is affected



NATIONAL GEOGRAPHIC
Photograph by Davide Lopresti

YOUR SHOT, THE WEEKLY WRAPPER
© COPYRIGHT DAVIDE LOPRESTI. ALL RIGHTS RESERVED.

Large increase in diatom productivity on slides, on rock and on sediment.
Not much increase in cyanobacteria until extremely high CO₂ levels



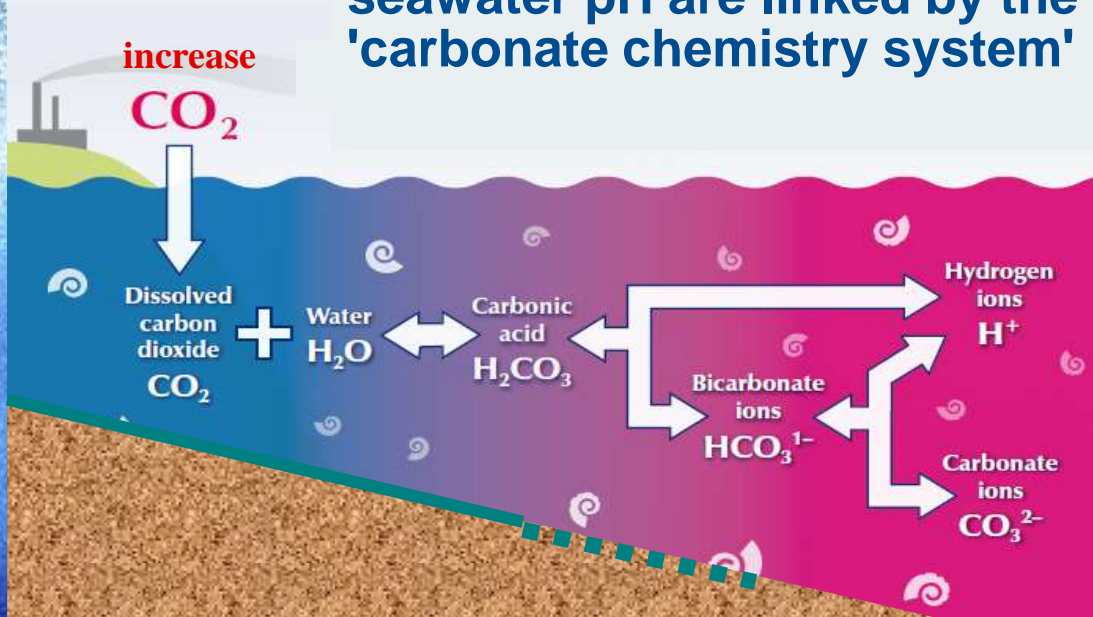
Johnson VR et al. (2015)
J. Mar. Sci. Eng. 3, 1425-1447

Recruitment from the plankton severely disrupted at high CO₂



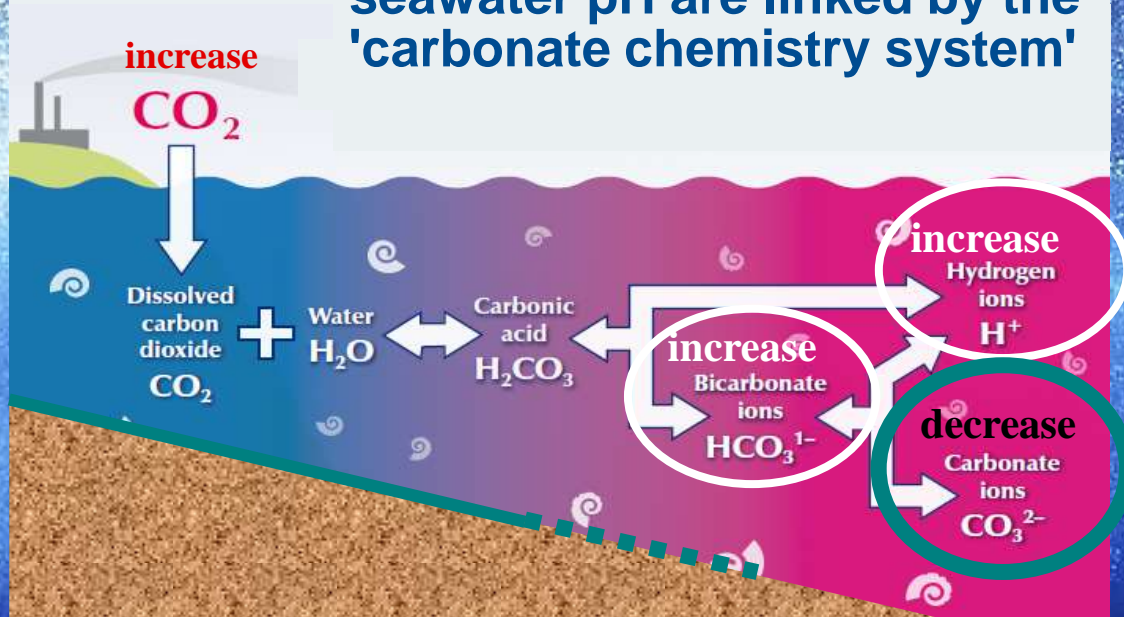
Cigliano et al. (2010) Marine Biology,
Smith et al. (2016) Nature Climate Change
Allen et al. (2017) Mar. Poll. Bull.

Atmospheric CO₂ and seawater pH are linked by the 'carbonate chemistry system'



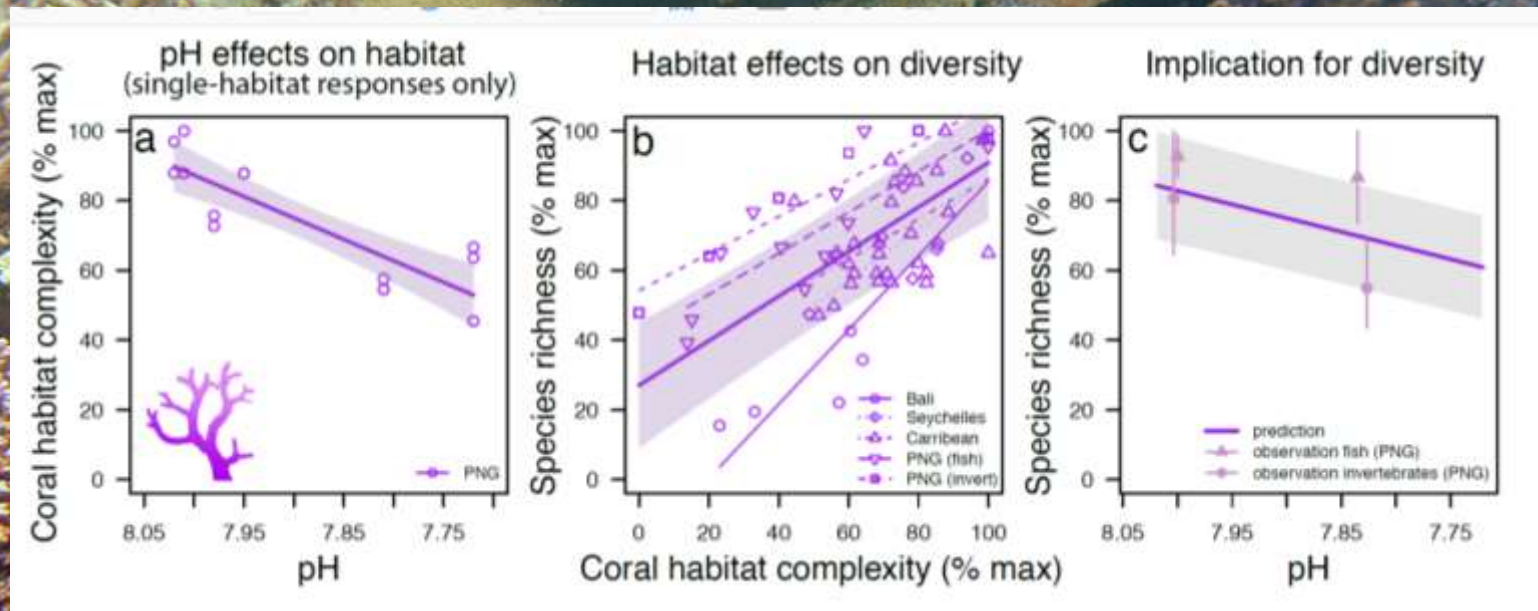
Ocean acidification is not just pH change
It is a multiple stressor

Atmospheric CO₂ and seawater pH are linked by the 'carbonate chemistry system'



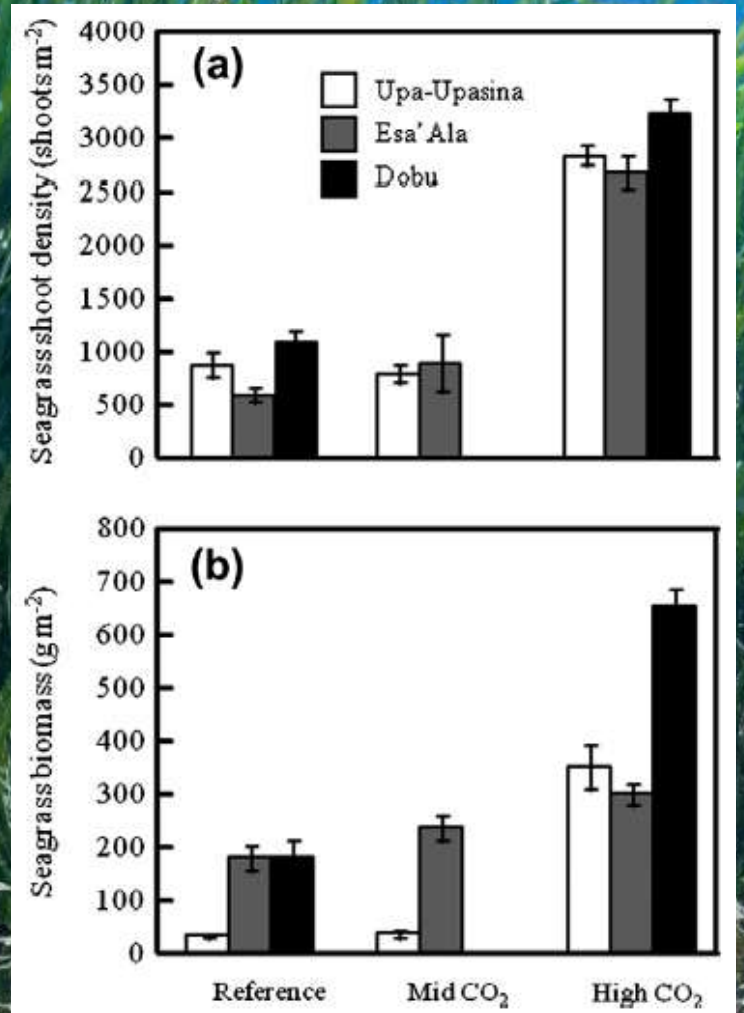
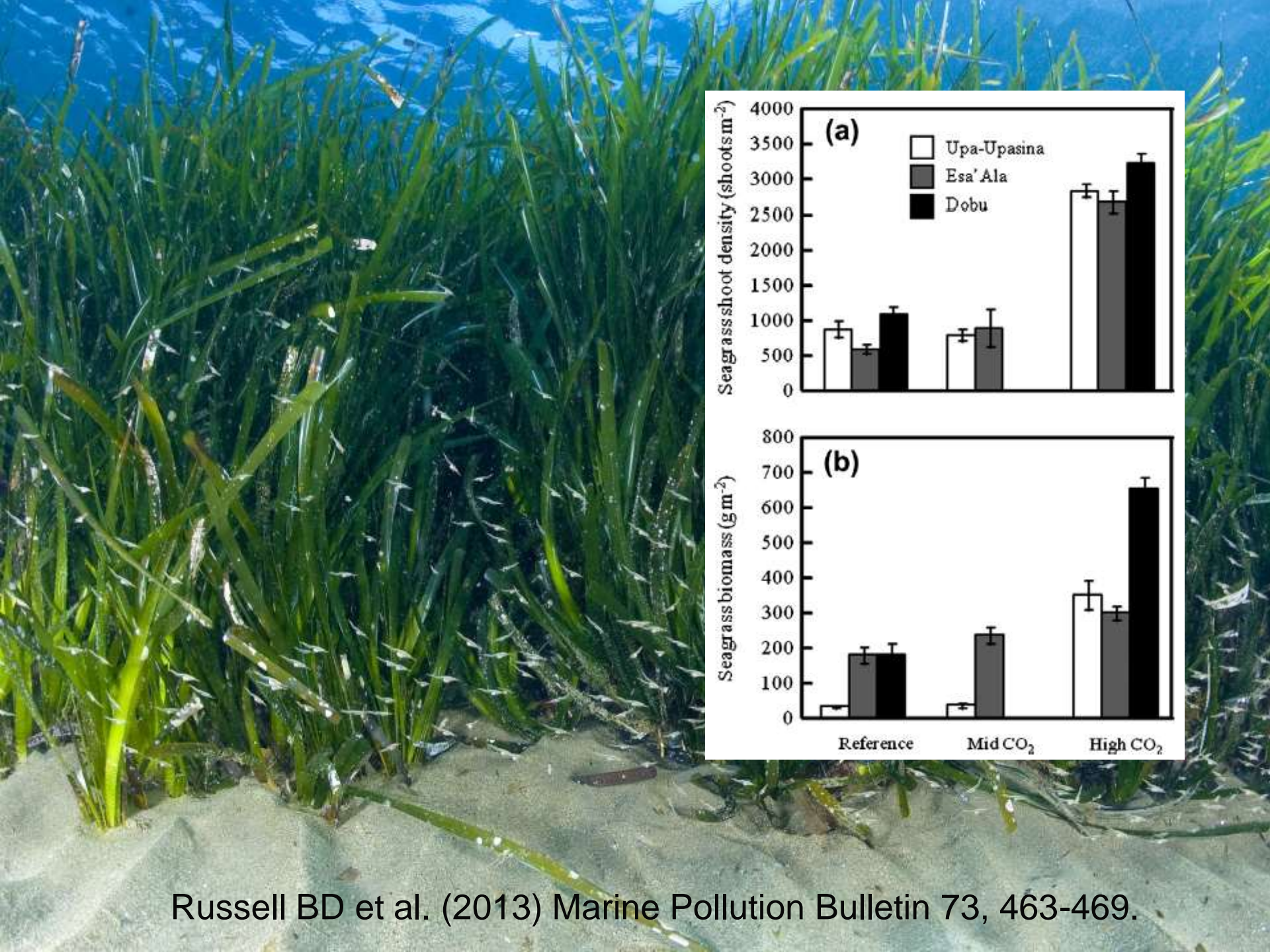
Increased DIC can be a resource for primary producers. Decreased carbonate saturation can corrode shells and skeletons.

OA mediates biodiversity shifts via biogenic habitat modifications

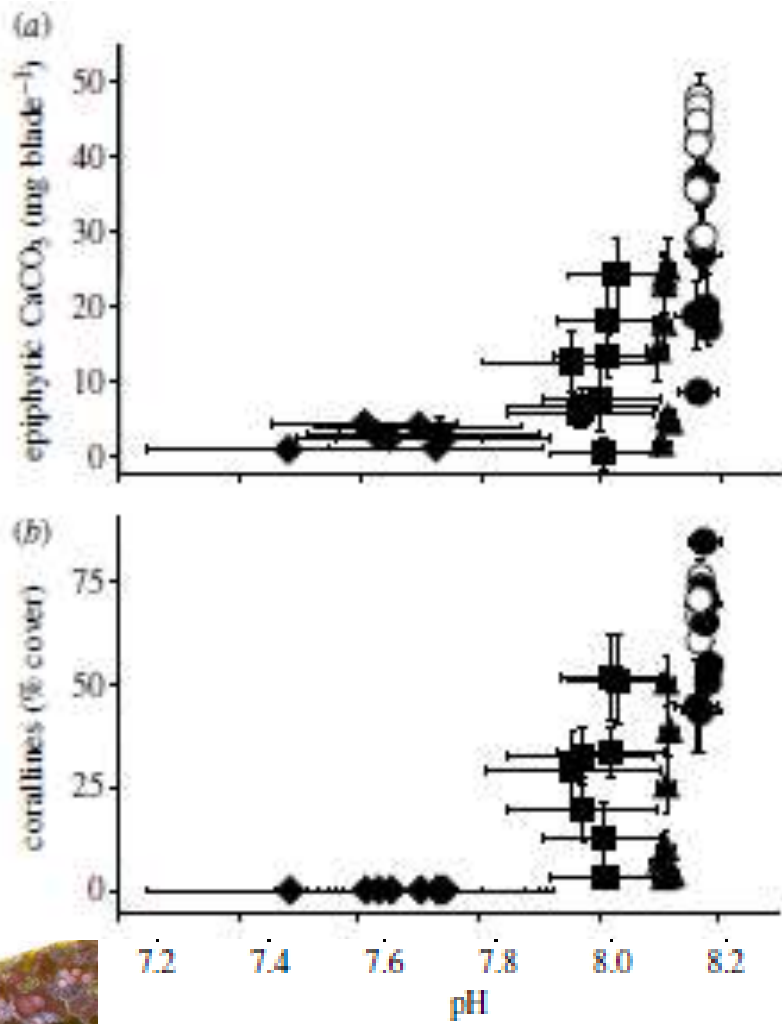




Seagrass is carbon limited; it grows well at the seeps



Russell BD et al. (2013) Marine Pollution Bulletin 73, 463-469.



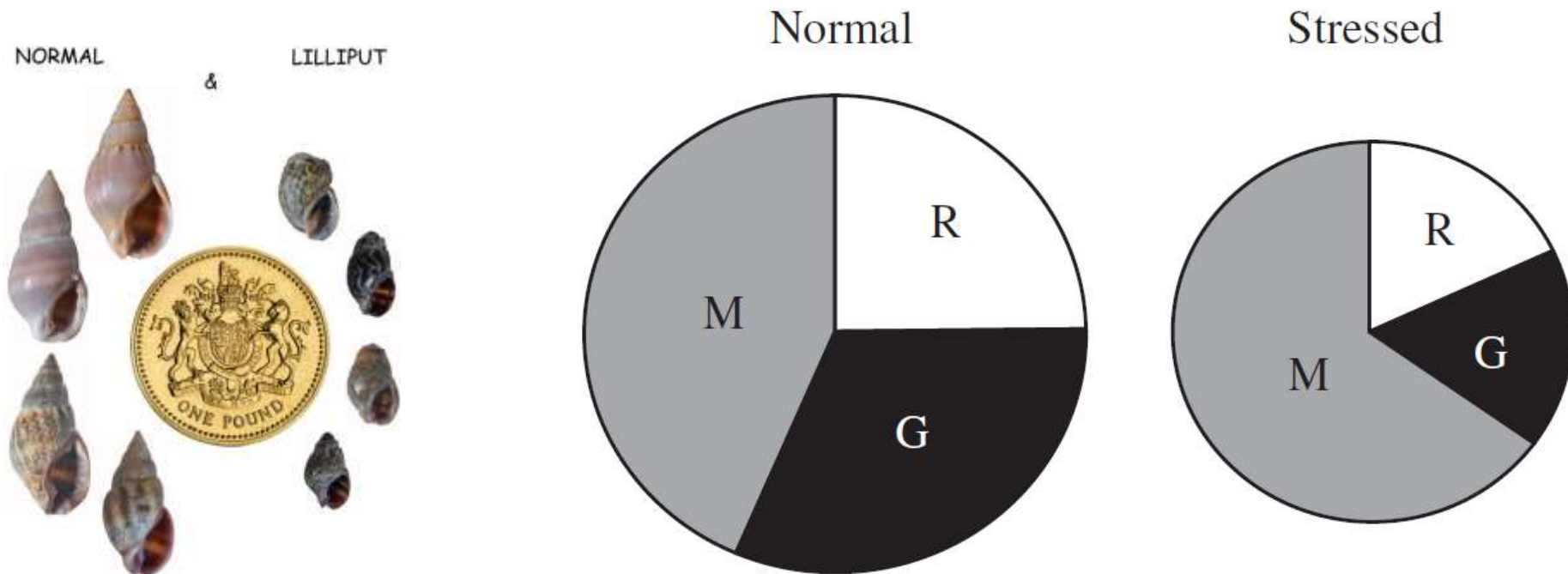
Martin et al. (2008) *Biology Letters*

We used $\delta^{13}\text{C}$ values as a proxy for $\text{CO}_2:\text{HCO}_3^-$ use



Porzio et al. 2010 J Exp Mar Biol Ecol
Cornwall et al. (2017) Scientific Reports

Physiological advantages of dwarfing in surviving extinctions in high-CO₂ oceans



Metabolism, Reproduction and Growth are adversely affected by stress

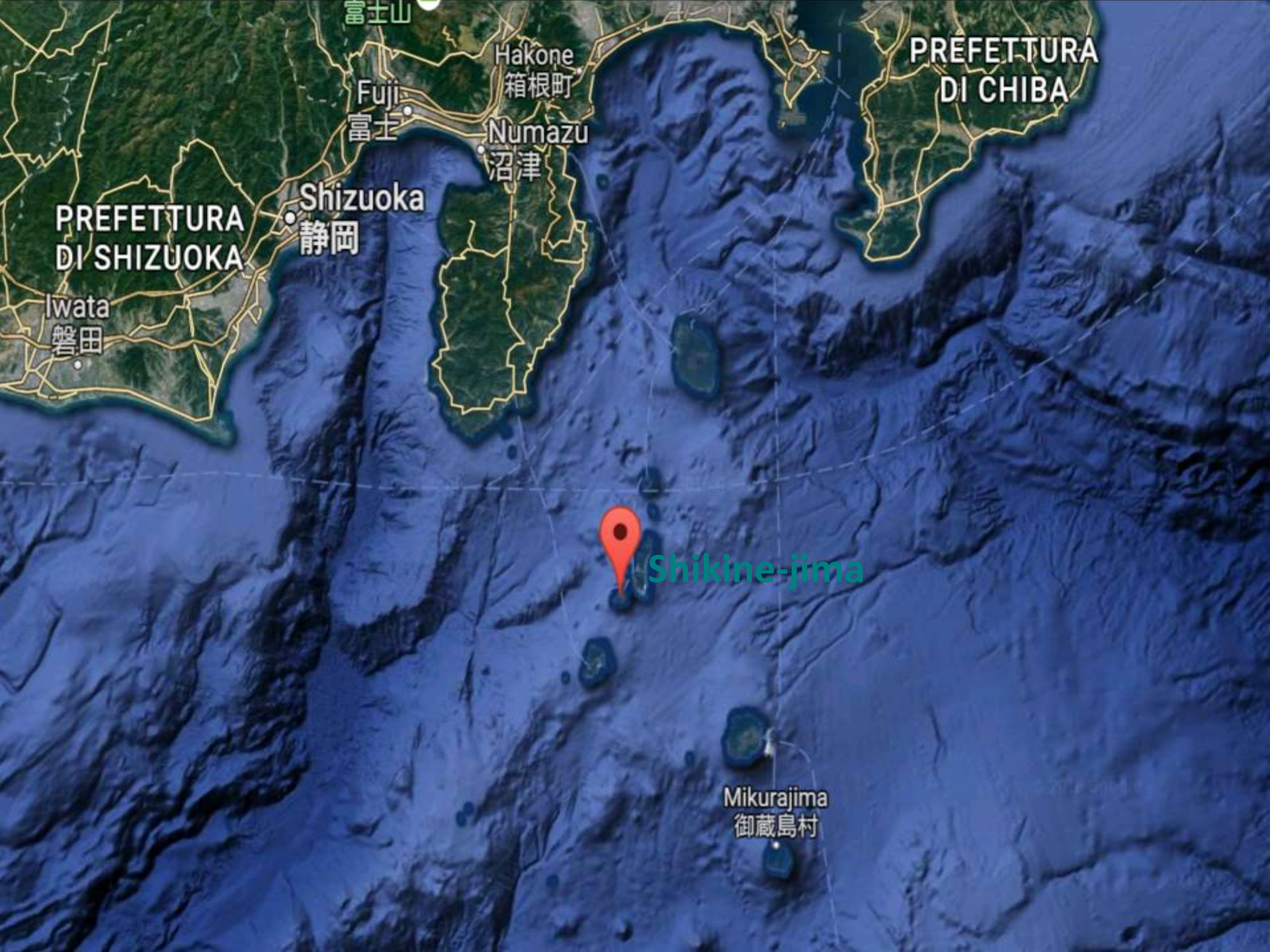
High CO₂



High CO₂ plus warming



**Warming makes many organisms more susceptible to ocean acidification
(Rodolfo-Metalpa et al. 2011 Nature Climate Change)**



富士山

Hakone
箱根町

Fuji
富士

Numazu
沼津

PREFETTURA
DI CHIBA

PREFETTURA
DI SHIZUOKA
Shizuoka
静岡

Iwata
磐田

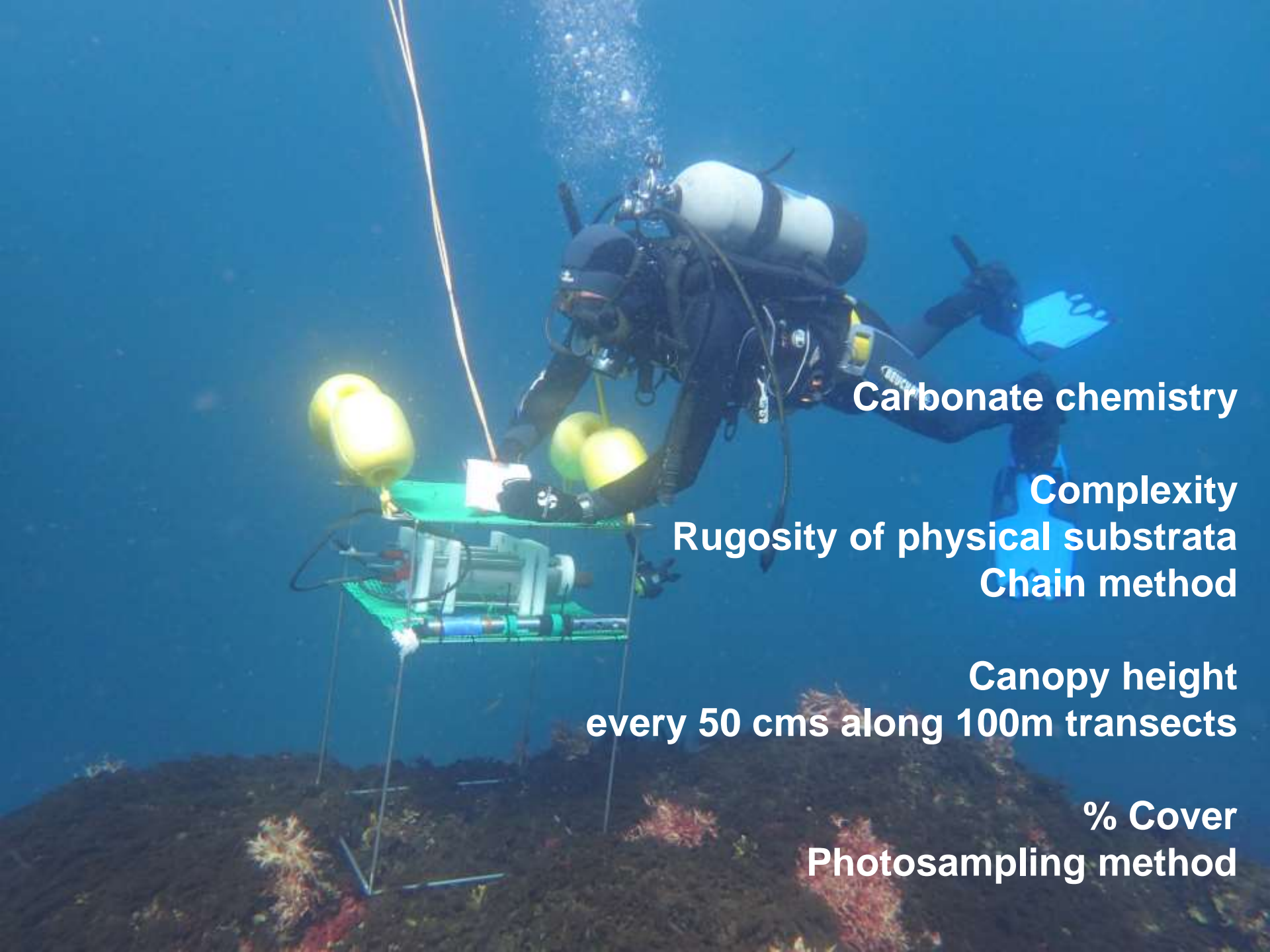
Shikine-jima

Mikurajima
御蔵島村

CO₂ seeps at Shikine-Jima

Agostini et al. (2018) Sci Rep





Carbonate chemistry

Complexity

Rugosity of physical substrata

Chain method

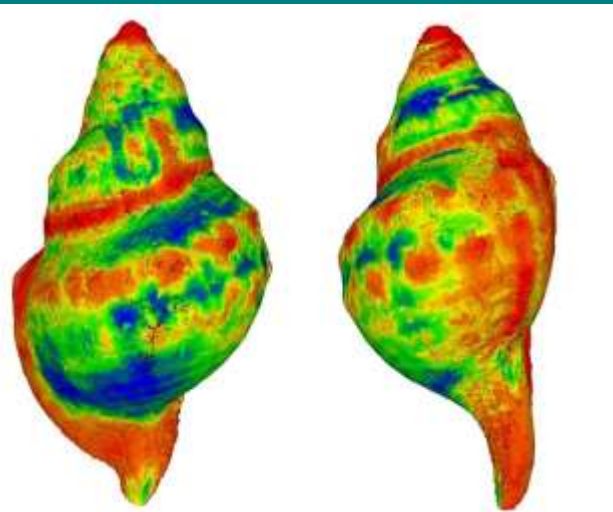
Canopy height

every 50 cms along 100m transects

% Cover

Photosampling method

Using computed tomography (CT) scanning, we measured the thickness, density and structure of the shells. Shell thickness was halved in areas with raised CO₂ while average shell length was reduced from 178 mm in sites with present day levels to 112 mm. In some cases, body tissue exposed, with the corrosive effects of acidification far more pronounced around the oldest parts of the shell.

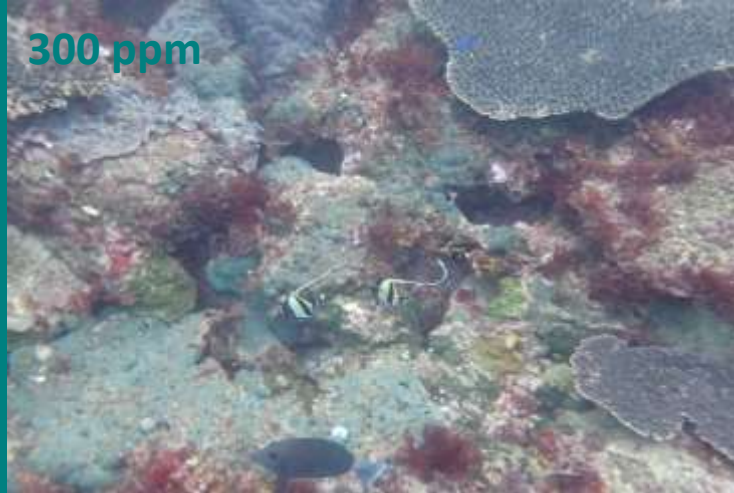


Harvey et al. 2018
*Frontiers in Marine
Science*

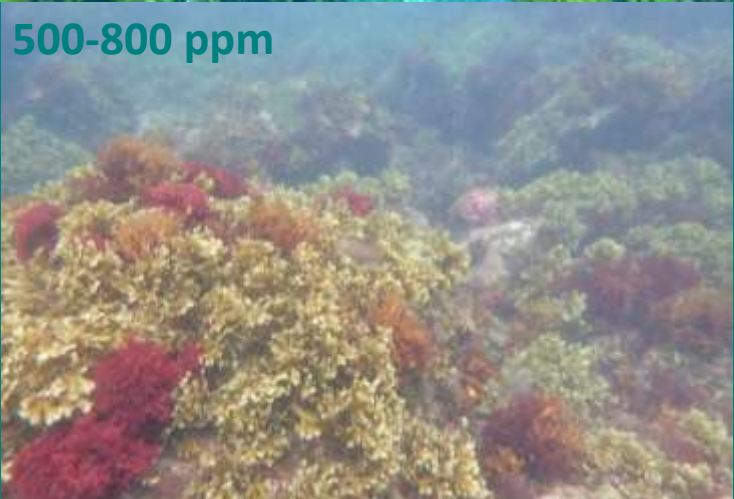
300 ppm



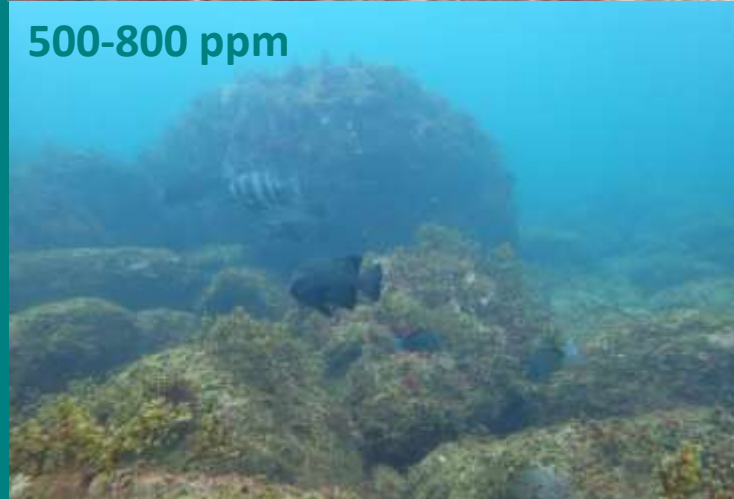
300 ppm



500-800 ppm



500-800 ppm



>900 ppm



>900 ppm



May 2017

September 2017



Present day CO₂ levels

An underwater photograph showing a dense carpet of green seagrass and brown algae on the seabed. The seagrass is the dominant feature, covering most of the bottom. There are patches of brown algae and some yellowish, possibly dead or bleached, seagrass. A small blue object is visible in the middle ground. The overall scene suggests a healthy but potentially stressed marine ecosystem.

CO₂ levels 900 ppm in 2090 BAU

Fish assemblages are affected by biogenic habitat shifts

Underwater Visual Censuses (UVC)

3 sites along the gradient (from 1200 to 500 ppm) and 2 ambient sites (300 ppm)

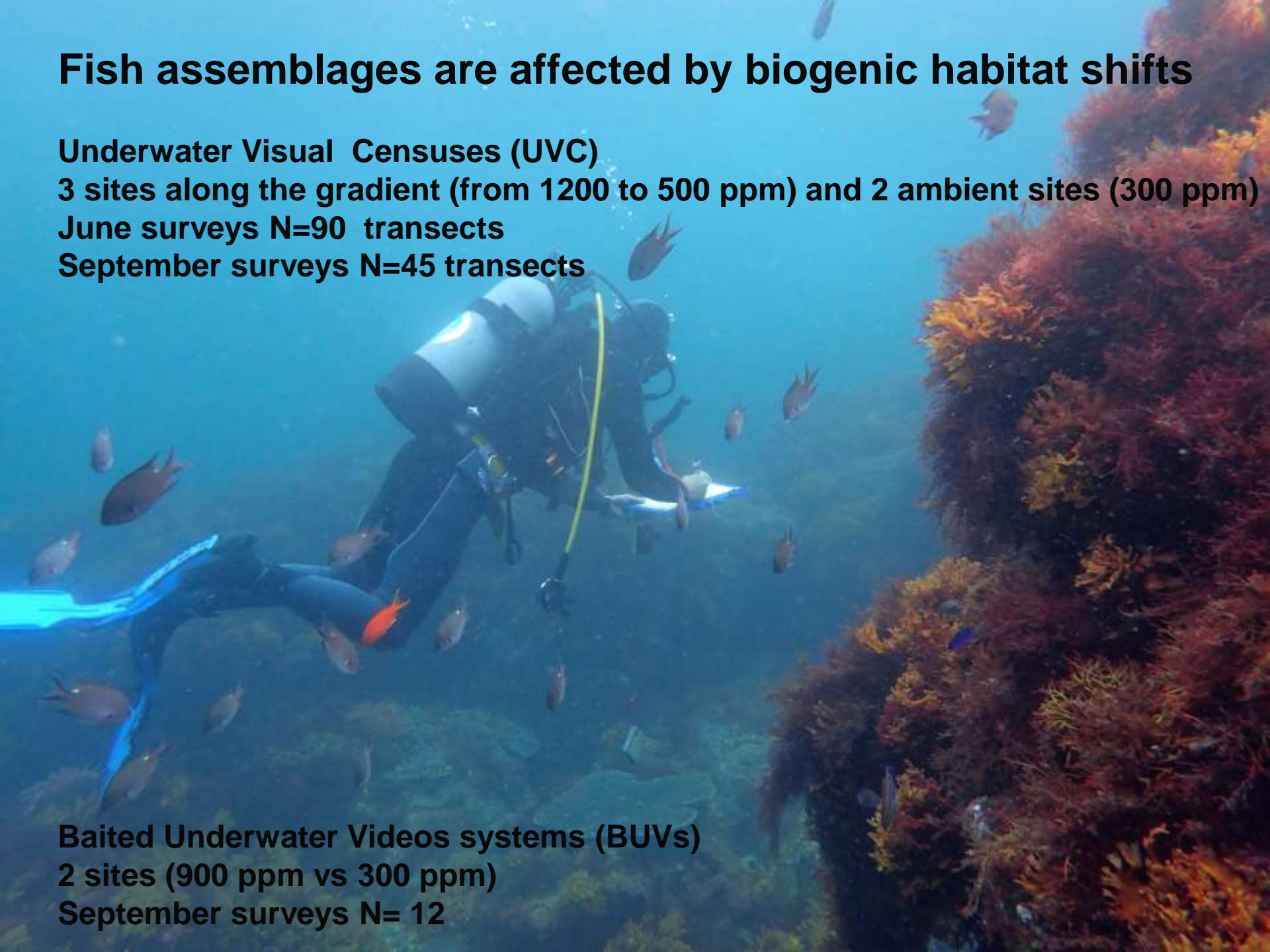
June surveys N=90 transects

September surveys N=45 transects

Baited Underwater Videos systems (BUVs)

2 sites (900 ppm vs 300 ppm)

September surveys N= 12



Ambient CO₂

↑ Canopy
↑ Corals

↑↑ Fish diversity
↑ Fish abundance



900 ppm CO₂
In 2090 if follow BAU

Total of ca. 70 fish spp.

94% in Ambient CO₂
only
56% in Elevated CO₂



Current cuts are not enough!

CO ₂ emissions	Δ sea surface temperature (°C)	Δ surface ocean pH
Present day	0.83	-0.11
RCP 2.6	1.13	-0.15
NDCs*	2.0 to 2.6	-0.26 to -0.32
RCP8.5	3.15	-0.41

Paris World

NDCs

BAU

Global mean values in 2090-2099 relative to 1870-1899

*Nationally Determined Contributions from Climate Action Tracker and Climate Interactive

Amended from Magnan et al (2016) Nature Climate Change

OCEAN ACIDIFICATION IMPACTS ON COASTAL COMMUNITIES

Summary for policymakers from the Third International Workshop



"Bridging the Gap between Ocean Acidification and Economic Valuation"

OCEANOGRAPHIC MUSEUM,
PRINCIPALITY OF MONACO

12-14 January 2015



Can we produce a 4-page information pamphlet, tailored to Japanese and/or Asian stakeholders?

MAIN REPORT

OAI RUG
Ocean Acidification International
Research Group

ACTING ON OCEAN ACIDIFICATION
Improving prospects by planning ahead

Ocean acidification impacts are already happening. They are progressive. How rapidly these changes take hold all depends on the scale of emission cuts. Whilst cuts are needed urgently, we also now need to plan ahead. We need to invest for the future. This plan is designed to help us do just that.

20 FACTS about Ocean Acidification

November 2011

This document presents the highlights of the *Frequently Asked Questions* about Ocean Acidification (2010, 2012; www.whoi.edu/OCB-OA/FAQs), a detailed summary of the state of ocean acidification research and understanding. The FAQs and this fact sheet are intended to help scientists, science communicators, and science policy advisors asked to comment on details about ocean acidification. In all, 63 scientists from 47 institutions and 12 countries participated in writing the FAQ, which was produced by the Ocean Carbon and Biogeochemistry Project (www.us-ocb.org), the United Kingdom Ocean Acidification Programme (www.oceanacidification.org.uk), and the European Project on Ocean Acidification (EPOCA). More information and contacts can be found at any of these websites or at the Ocean Acidification International Coordination Centre's website (www.iaea.org/ocean-acidification). The Intergovernmental Panel on Climate Change (IPCC) Fifth Assessment Report findings on ocean acidification can be viewed at www.ipcc.ch.

1 Ocean acidification (OA) is a progressive increase in the acidity of the ocean over an extended period, typically decades or longer, which is caused primarily by uptake of carbon dioxide (CO₂) from the atmosphere. It can also be caused or exacerbated by other chemical additions to the ocean. OA can be more severe near where human activities release CO₂, such as acid rain and coal-burned power plants.

2 OA has been well documented with both observations and model simulations over several decades by hundreds of researchers. OA has been definitively linked to human-generated CO₂ in the atmosphere that has been released primarily from fossil fuel combustion and land use changes.

3 Acidity may be thought of as simply the hydrogen ion concentration (H⁺) in a liquid, and pH is the logarithmic scale on which this concentration is measured. It is important to note that acidity increases as the pH decreases.

4 Average global surface ocean pH has already fallen from a pre-industrial value of 8.2 to 8.1, corresponding to an increase in acidity of about 30%. Values of 7.8–7.9 are expected by 2100, representing a doubling of acidity.

5 The pH of the open-ocean surface layer is unlikely to ever become acidic (i.e. drop below pH 7.0), because seawater is buffered by dissolved salts. The term "acidification" refers to a pH shift towards the acidic end of the pH scale, similar to the way we describe an increase in temperature from -20°C to 0°C (-4°F to 32°F); it's still cold, but we say it's "warming."

6 OA is also changing seawater carbonate chemistry. The concentrations of dissolved CO₂, hydrogen ions, and bicarbonate ions are increasing, and the concentration of carbonate ions is decreasing.

7 Changes in pH and carbonate chemistry force marine organisms to spend more energy regulating chemistry in their cells. For some organisms, this may have less energy for other biological processes like growing, reproducing or expanding to other areas.

Fact sheet also called Sea Surface Acidity, an early special interest group of the International Ocean Acidification Programme (IOAGI).

OCEAN ACIDIFICATION
Summary for Policymakers
Third Symposium on the Ocean in a High-CO₂ World

SCOR **GLOBAL IGBP CHANGE**