Ocean Acidification: A Global Problem With Local Impacts

Richard A. Feely Pacific Marine Environmental Laboratory/NOAA Seattle WA 98037

 Outline
 Physical/Chemical Oceanography
 Marine Biology Including:

 Phytoplankton,
 Zooplankton (oyster larvae pteropods) and Salmon

Ocean Policy Research Institute Sasakawa Peace Foundation International Symposium on Ocean Acidification 28 October 2018







Photo credit: Meghan Shea

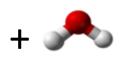
Ocean Acidification: the other CO₂ problem

Sarah R. Cooley (scooley@whoi.edu)

Climate change **Carbon Dioxide** (CO_2) CO_2 is an acid gas... it reacts with water to Ocean acidification form carbonic acid.

$CO_2 + H_2O \leftrightarrow H_2CO_3 \leftrightarrow HCO_3^- + H^+$

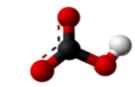




water



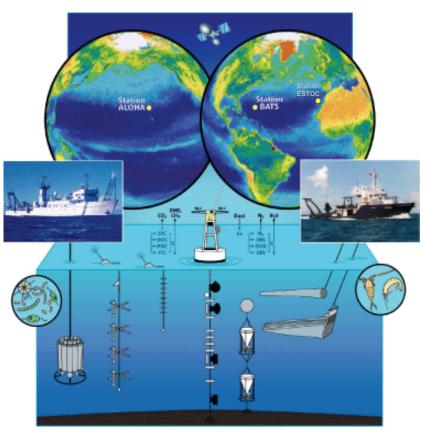
carbonic acid



bicarbonate ion

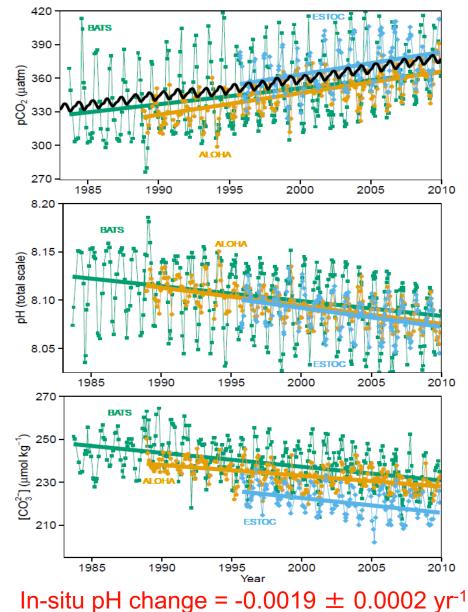
hydrogen ion

Change in pH from ocean acidification already measurable



<u>Data</u>: *Bates (2007) Dore et al. (2009) Santana-Casiano et al. (2007) Gonzàles-Dàvila et al. (2010)*

IPCC AR5 WG1 Report, Chap. 3 (2013)

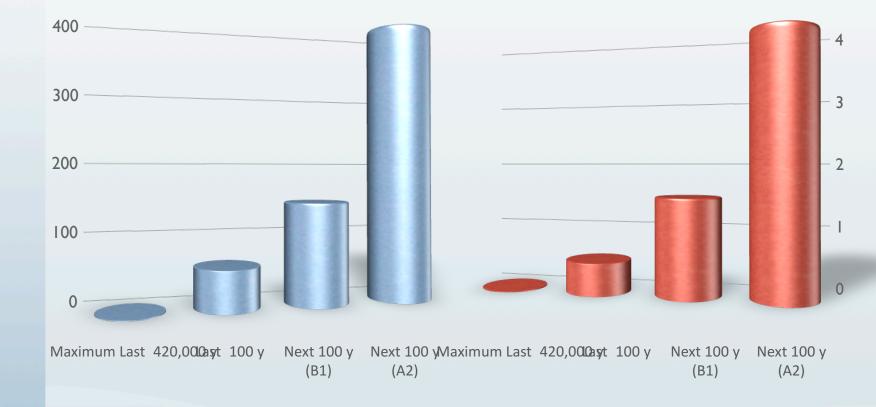


Rates of increase are important

Atmospheric CO₂ Rate of rise in CO₂ (ppm/100y)

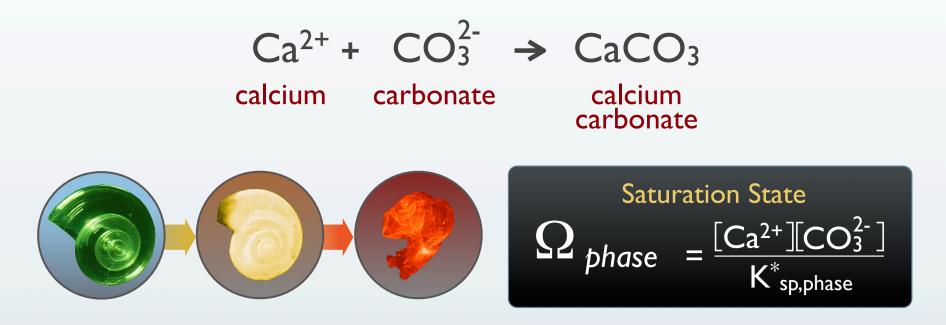
Global Temperature

Rate of rise in global temperature (°C/100)



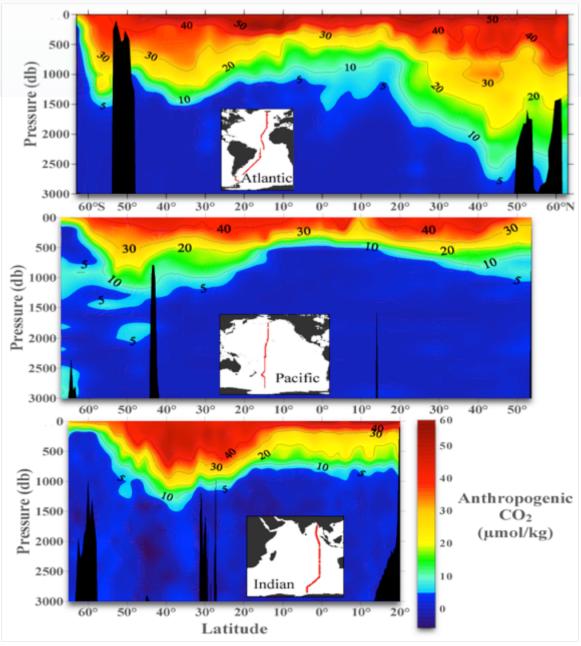
Hoegh-Guldberg et al. 2007, Science

Saturation State



- $\Omega > I$ CaCO₃ precipitates
- Ω =1 equilibrium
- $\Omega < I$ CaCO₃ dissolves

Common carbonate minerals: aragonite (more soluble) and calcite (less soluble)

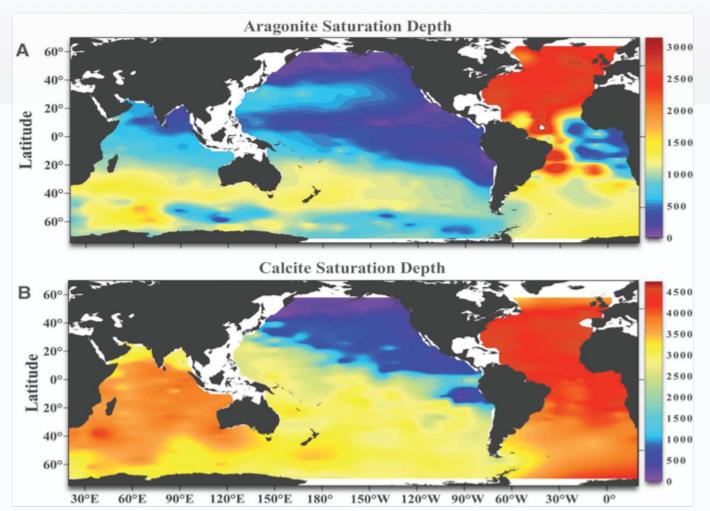


Penetration of Anthropogenic CO₂ into Ocean

- Difference of present-day levels minus pre-industrial (year 1800)
- Half trapped in upper 400 m
- Equivalent to about a third of all historical carbon emissions
- 150 Pg C since the beginning of the industrial era have accumulated in the oceans

Sabine et al. Science 2004

Observed aragonite & calcite saturation depths



The **aragonite saturation state** migrates towards the surface at the rate of 1-2 m yr⁻¹, depending on location.

Feely et al. (2004)

Surface ocean pH change since the industrial revolution

1850



Early vulnerabilities include polar and tropical oceans



IGBP IOC SCOR (2013) Ocean Acidification Summary for Policy Makers

Ocean Acidification is Occurring Rapidly

- Approximately 28% of the CO₂ generated by human activities since the mid-1700s has been absorbed by the oceans.
- Ocean acidity has increased 30% since the start of the industrial age.
- Ocean acidity is projected to increase 100-150% percent by 2100.
- Current rate of acidification is nearly 10x faster than any period over the past 50 million years.

Seasonal invasion of corrosive upwelled water on the west coast of North America

- Upwelling of CO₂-rich intermediate waters, undersaturated with aragonite (Ω_{arag}), onto continental shelf from a depth of 150 – 200m
- Exposure of productive coastal ecosystems to corrosive upwelled water

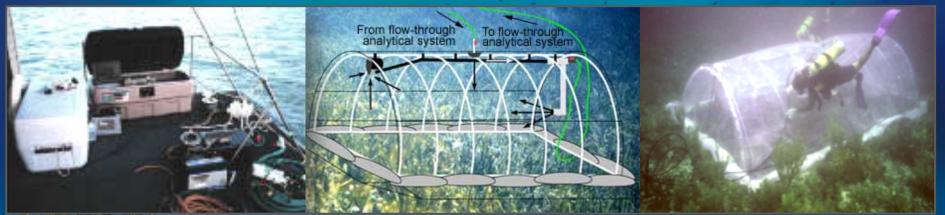
The Effects on Marine Organisms

Experiments on *many scales*



Biosphere 2 (provided by Mark Eakin)

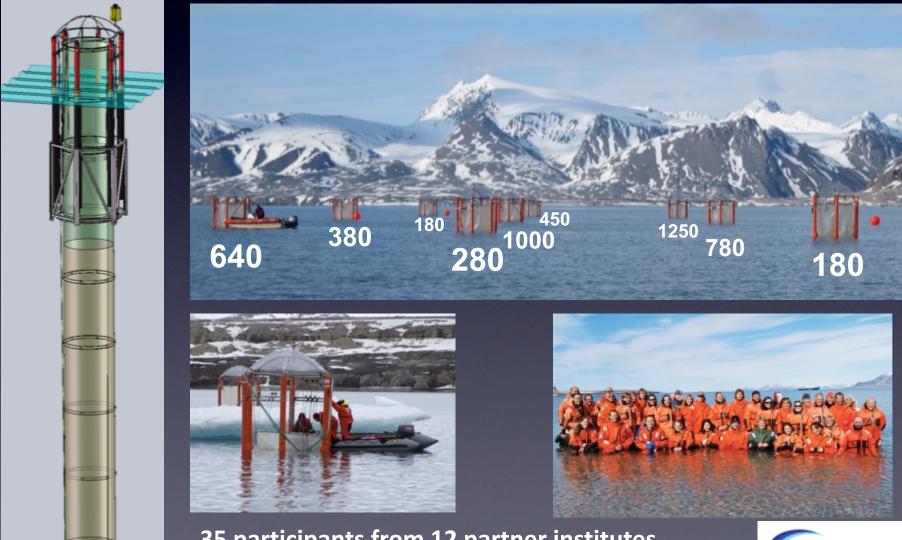
Aquaria & Small Mesocosms





SHARQ Submersible Habitat for Analyzing Reef Quality

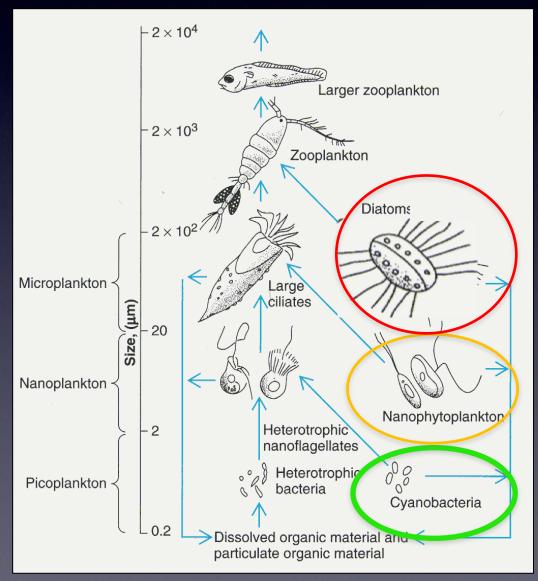
Svalbard 2010: CO₂ Enriched Mesocosms



35 participants from 12 partner institutes



Major Changes at the Base of the Pelagic Foodweb



(unpublished data coutesy Ulf Reibesell, BIOACID and EPOCA)

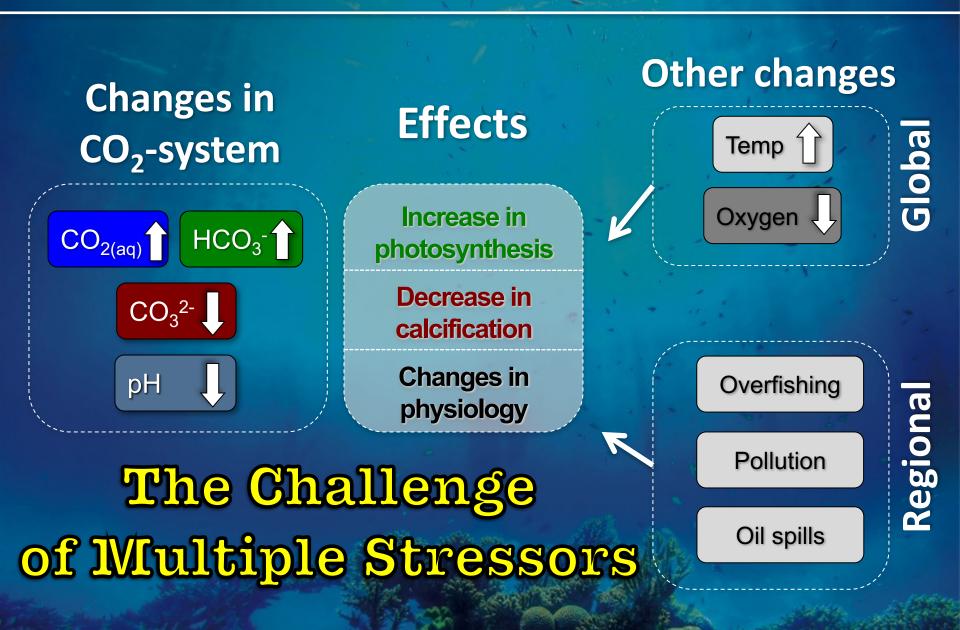
.... with likely consequences for higher trophic levels and foodwebs

outcompeted at high CO₂

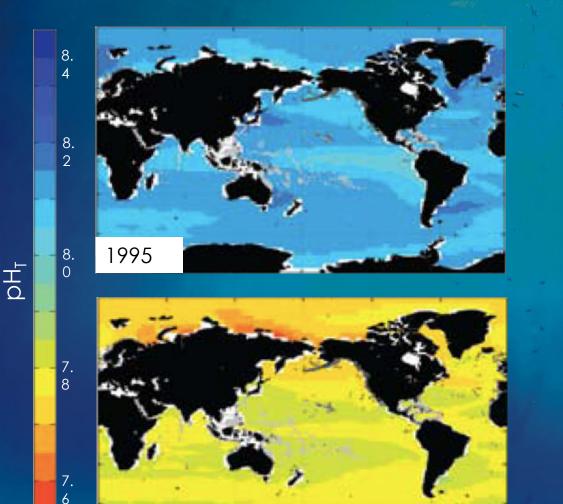
moderately stimulated at high CO₂

strongly stimulated at high CO₂

How CO₂ in seawater affects marine life



A meta-analytic approach



Study criteria:

 $\Delta \text{ pH} \le 0.5$ decrease

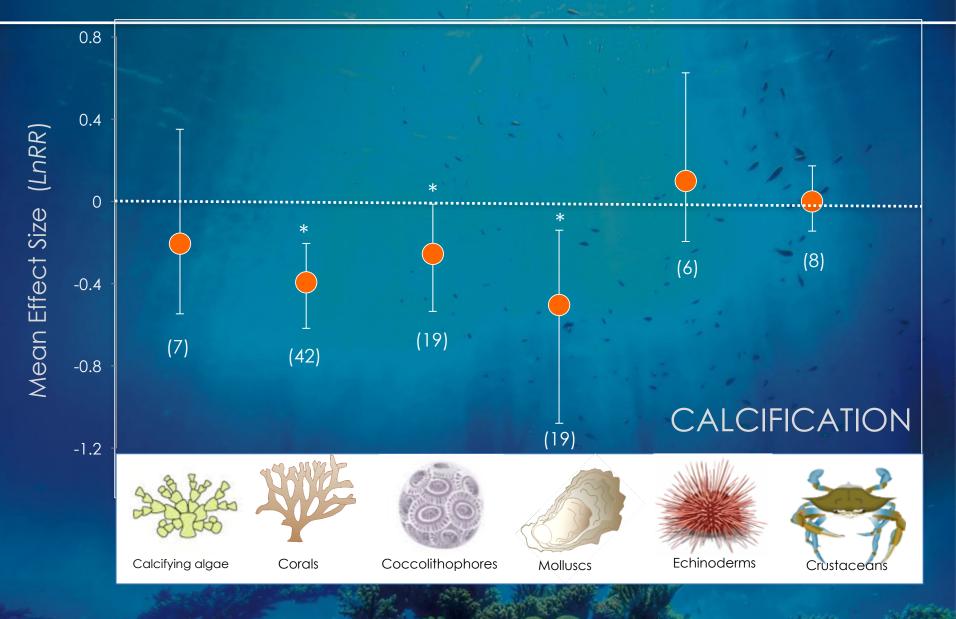
588 unique experiments

Kroeker et al., 2013

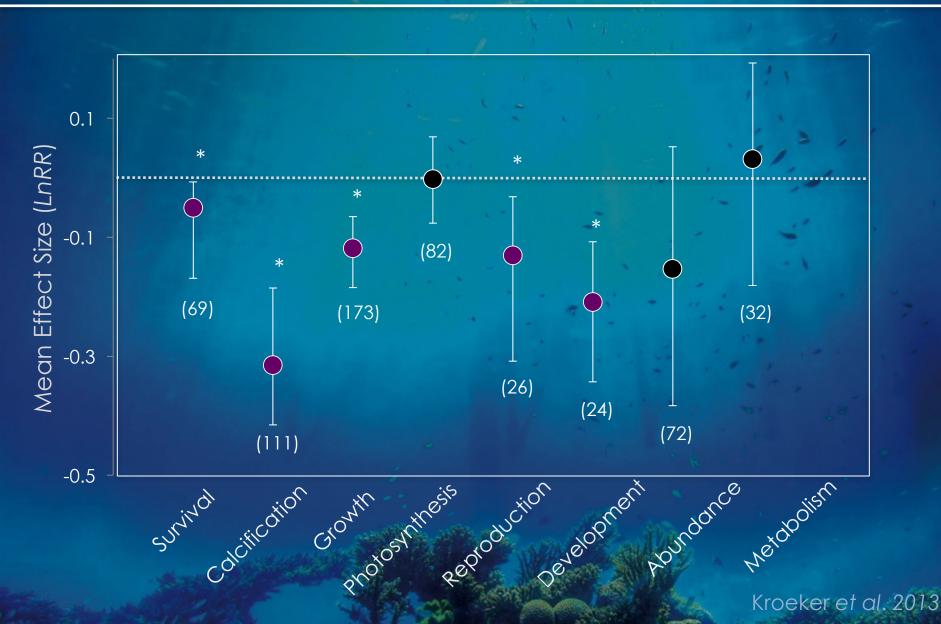
CC SM3-modeled decadal mean pH (surface)

2095

Variation in sensitivity among calcifiers



OA has negative effects across a large range of processes



Effects on Ecosystems

A window into the future of coral reefs?

pH 8.05: Today pH 7.95: ~ year 2050 pH 7.8: ~ year 2100

Fabricius et al. (2010) Nat Clim Change

Slide courtesy Katharina Fabricius

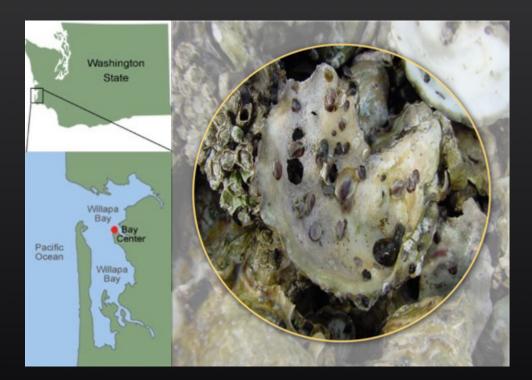
Effects on Calcification

Many other species



Pacific Northwest *oyster emergency*

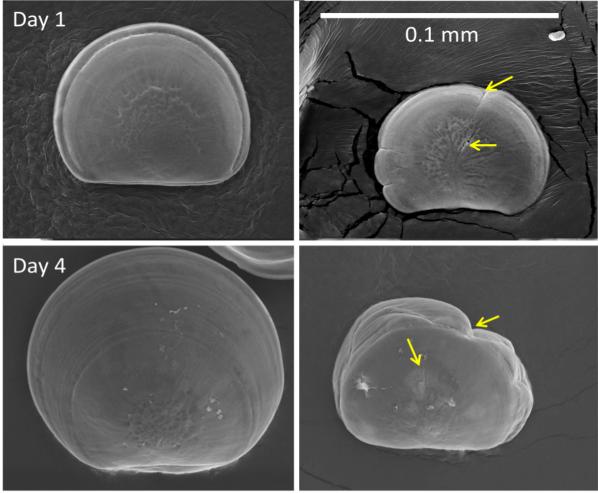
Are larval oysters the "canary in the coal mine" for near-shore acidification?



West Coast Oyster seed crisis

- Failure of larval oyster recruitments in recent years
- Commercial oyster hatchery failures threatens a \$110M industry (3200 Jobs)
- Low pH "upwelled" waters a possible leading factor in failures

Oyster Larvae in Peril



 Ω Aragonite = 1.64 pCO2 = 403 ppm pH (total) = 8.00

 Ω Aragonite = 0.47 pCO2 = 1418 ppm pH (total) = 7.49

Larvae reared in low aragonite saturation state conditions appear able to form their shell, but suffer from major defects that ultimately result in mortality.

Figure source: Elizabeth Brunner and George Waldbusser, Oregon State University

Pteropod Dissolution as an indicator of past, present and future impacts

Pre-industrial level of dissolution due only to upwelling: naturally occurring dissolution (~18%)

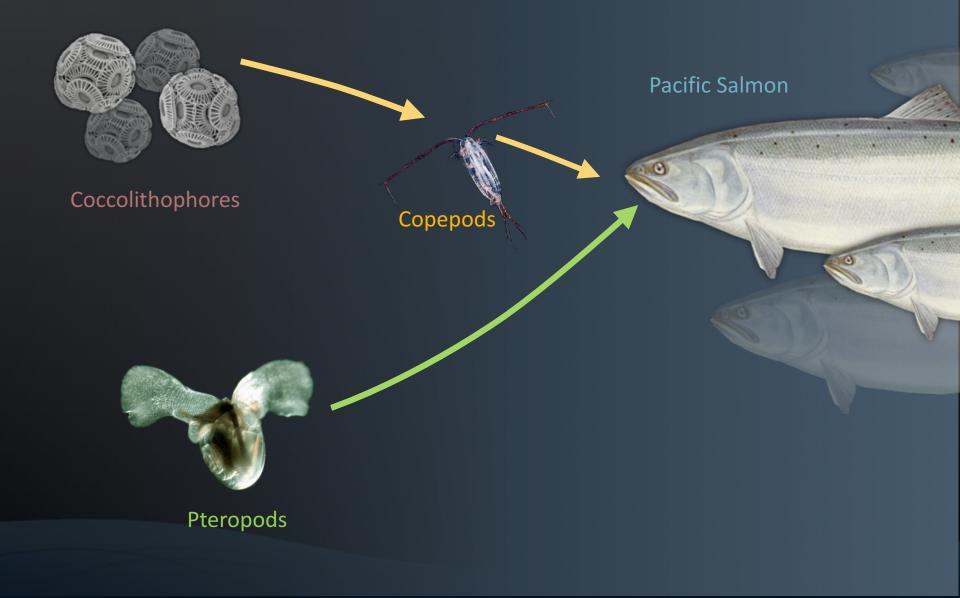


Significant increase in the current level of dissolution \rightarrow 53% in the coastal regions with anthropogenic CO₂ contribution in addition to upwelling.

By 2050: ~70% of water column will be undersaturated \rightarrow 70% of pteropods affected by severe dissolution in the coastal regions.

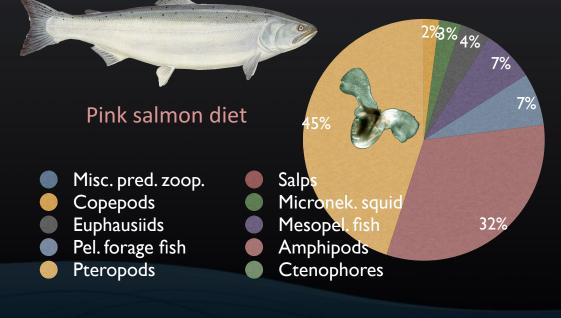
Bednarsek et al 2014

Potential *food web* impacts



Marine Fish impacts

Western Alaskan Sockeye
 British Columbia Sockeye
 Central Alaskan Pink
 Japanese Chum

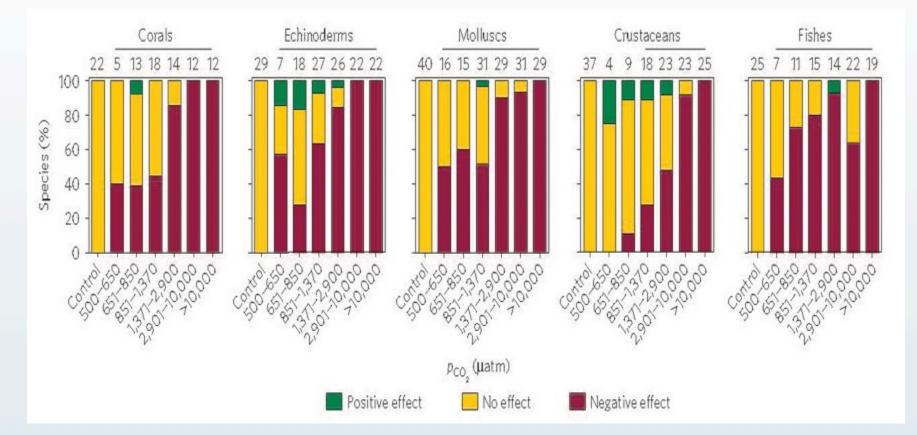


Predicted effect of climate change on pink salmon growth:

- 10% increase in water temperature leads to 3% drop in mature salmon body weight (physiological effect).
- 10% decrease in pteropod production leads to 20% drop in mature salmon body weight (prey limitation).

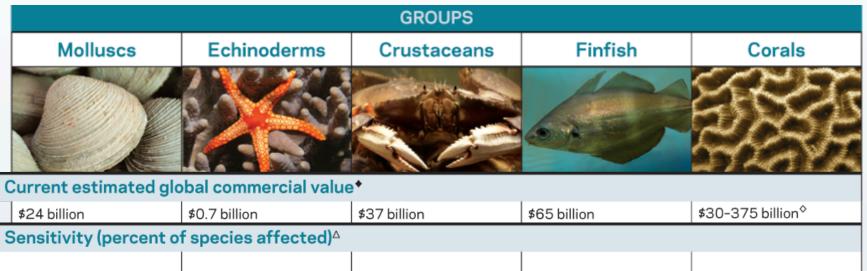
Aydin et al., 2005

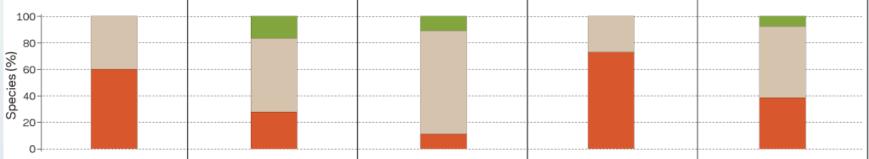
Impacts of Acidification (after Wittmann and Pörtner, 2013)



Corals, echinoderms and molluscs are more sensitive to 936 ppm pCO₂ than are crustaceans. Larval fishes may be even more sensitive than the lower invertebrates, but taxon sensitivity on evolutionary timescales remains obscure. The variety of responses within and between taxa, together with observations in mesocosms and palaeo-analogues, suggest that ocean acidification is a driver for substantial change in ocean ecosystems this century, potentially leading to long-term shifts in species composition.

Commercially Important Organisms





Effects Positive None Negative

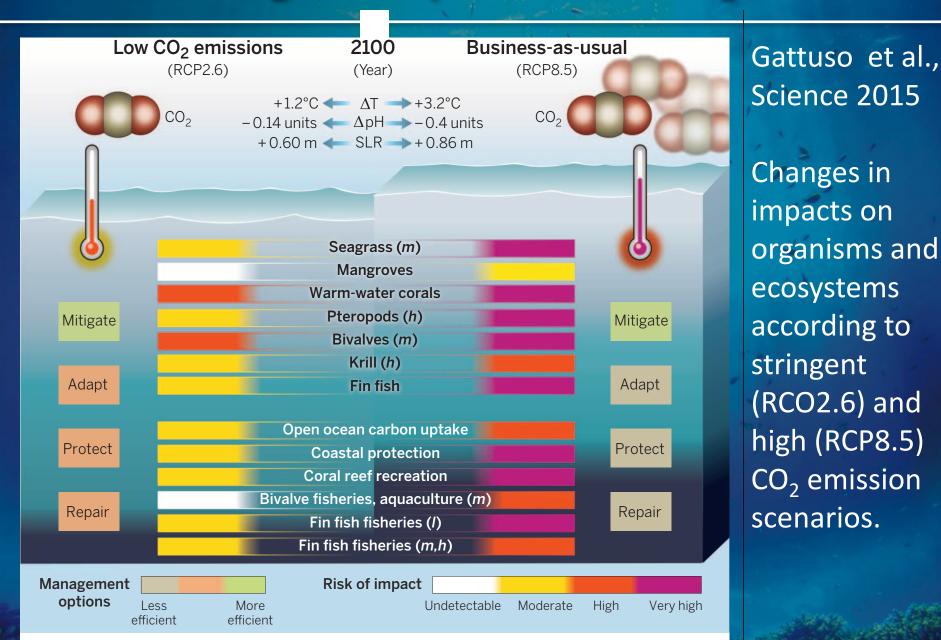


Concern for Marine Organisms and Ecosystems

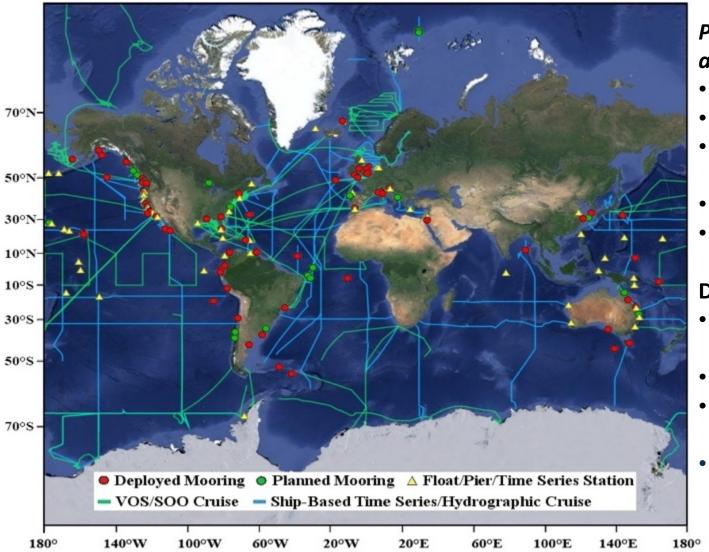


- Reduced calcification rates
 Significant shift in key nutrient and trace element speciation
 Shift in phytoplankton diversity
 Reduced growth, production and life span
- of adults, juveniles & larvae
 Reduced tolerance to other environmental fluctuations
- Changes to fitness and survival
- Changes to species biogeography
- Changes to key biogeochemical cycles
- Changes to food webs
 - Reduced sound absorption
- Reduced homing ability
- Reduced recruitment and settlement
- Changes to ecosystem goods & services
- **Changes to behavior responses**

Future Biological Impacts from CO₂ Emissions



Global Ocean Acidification Observing Network



GOA-ON

Global Ocean Acidification Observing Network

Provide

access to:

- SOCAT SOOP
- Data
- GLODAPv2 Repeat Hydrography data
- Mooring data
- Coastal Data

Develop

- seasonal and trend views
- synthesis products
- model output
- <u>www.GOA-ON.org</u>

Conclusions



> The ocean is acidifying rapidly Some species will be sensitive to OA Biological responses to OA are variable Impacts of OA can transfer through food webs Other stressors can exacerbate response to OA Economic consequences of OA are significant

Source: L. Whitely Binder, CIG